

Chichibabin-Type Condensation of Cyclic Ketones with 3-R-1,2,4triazin-5(4H)-ones

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Supporting Information

ABSTRACT: Reactions between substituted 1,2,4-triazines and ketones were investigated. General procedures for one-pot synthesis of hydrogenated derivatives of such polycyclic systems as benzo[c][1,2,4]triazino[1,6-a][2]-azecine, [1,2,4]triazino[1,6-f]phenantridine, and dicyclopenta[b,d]pyrido[1,2f][1,2,4]triazine are described.

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INTRODUCTION

Fused pyridines are widespread among natural organic compounds; they are of great importance for medicine. 2 Pyridines, fused with two aliphatic rings, have drawn considerable attention of researchers, as 1,2,3,4,7,8,9,10octahydrophenantrene shows pesticide activity.3 Triazinefused pyridines are known to exhibit antidepressant activity.⁴ The most popular synthetic route to bis-fused angular pyridines is Chichibabin's synthesis from cyclic ketones, ammonia, and aldehydes. The reaction requires high pressure^{3,5} or hydrothermal conditions in aqueous ammonium chloride.⁶ Procedures that utilize activated forms of cyclic ketones, such as the product of dimerization of cyclohexanone-1,1'-bi-(cyclohexilidene)-2-one, enamines of cyclic ketones, 2b,8 are noteworthy. Syntheses from pyrylium salts⁹ and cyclobutadienes¹⁰ are also known. The source of nitrogen in the molecule of fused pyridine in most cases is ammonia, 5,6,9 while amides,^{7a} urea,^{7b,8d} 2-azadienes,^{8a} enamines,^{8b} carbodiimides,^{8c} ethylcyanoacetate,¹⁰ and dienamides¹¹ are also used.

■ RESULTS AND DISCUSSION

In this paper, we describe a detailed study of synthetic methods for synthesis of 1,2,4-triazine derivatives of angular bis-fused pyridines in the reactions of cyclic ketones with 1,2,4-triazines. Compounds containing fragments of 1,2,4-triazine (6-azauridine) arouse interest because some of them show anticancer, 12 antivirus, 13 and antibacterial activity. 14 It is known that 3-R-1,2,4-triazin-5(4*H*)-ones react with C-nucleophiles such as indoles, ¹⁵ pyrroles, ¹⁶ phenols, ¹⁶ and anilines. ¹⁶ We investigated the reaction between 3-R-1,2,4-triazin-5(4*H*)-ones and ketones. It was shown that the reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a,b) with acetophenone (2) in the presence of acids leads to

the formation of the expected products of nucleophilic addition to the double C=N bond of 3a,b (Scheme 1). When acetone was used, resinification of the reaction mixture was observed, while butanone-2 and hexanone-2 did not react.

Scheme 1

We previously demonstrated that cycloaddition of 3-R-1,2,4triazin-5(4H)-ones and cyclohexanone (4) or cyclopentanone (5) in the presence of trifluoroacetic acid leads to the formation of chichibabin-type fused pyridines.¹⁷ In the course of our research, we investigated the process of formation of fused pyridines. The search for proper conditions of the reaction of 3-R-1,2,4-triazin-5(4H)-one (1a) with cyclohexanone (4) under the action of bases showed only a methanolic solution of NaOCH₃ to be a suitable reaction medium. This reaction gives product 6a with low yield (7%). Thus, our studies were then switched to acidic conditions. The following conditions were modified to find an optimal combination: temperature, acid strength, and addition of some oxidative and dehydrating agents (Tables 1 and 2).

Irrespective of the mechanism, chichibabin synthesis of pyridines includes an oxidative stage; it is known that such

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6007

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Table 1. Reaction of 3-Ph-1,2,4-triazin-5(4H)-one (1a) with Cyclohexanone 4 under Different Conditions

acid	solvent	temp, °C	time	yield of 6a , %
p-TsOH	DMF	153	0.5 h	29
p-TsOH	DMF	153	12 h	49
p-TsOH	CH ₃ CN	82	1 h	34
CF ₃ COOH	DMF	100	6 h	30
CF ₃ SO ₃ H	DMF	100	6 h	22 ^a
AlCl ₃	DMF	20	1 day	36
p-TsOH, DDQ	CH ₃ CN	82	3 h	28
p-TsOH, DDQ, P ₂ O ₅	CH ₃ CN	82	3 h	21
CF ₃ COOH, CAN	DMF	20	3 days	19
CF ₃ COOH	DMF	20	7 days	44 ¹⁷
H_3PO_4	H_3PO_4	50	3 h	12
CH ₃ COOH	CH ₃ COOH	20	3 days	0
CH ₃ COOH	CH ₃ COOH	118	3 h	31
CF ₃ SO ₃ H	CH ₃ CN	82	6 h	31^a
CF ₃ SO ₃ H	CH ₃ OH	65	6 h	19 ^a
CF ₃ SO ₃ H	DMF	153	0.5 h	10 ^a
HCl	EtOH	20	1 day	40 ^b

^aYield of **6a** triflate (**6aa**). ^bYield of the **6a** chloride (**6ab**).

Table 2. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a-d) with Ketones 4 and 5 in the Presence of p-TsOH (1 equiv)

starting material	R	n	conditions	time, h	product	yield, %
1a	Ph	2	DMF	1	6a	44
1a	Ph	2	CH ₃ CN, P ₂ O ₅	3	6a	66
la	Ph	2	CH ₃ CN, mol sieves (3 Å)	3	6a	21
1b	SMe	2	DMF	1	6b	11
1b	SMe	2	CH ₃ CN, P ₂ O ₅	1	6b	0
1b	SMe	2	CH ₃ CN, P ₂ O ₅	3	6b	0
1c	4- MePh	2	DMF	1	6c	24
1c	4- MePh	2	CH_3CN , P_2O_5	3	6c	42
1d	4-ClPh	2	DMF	1	6d	15
1d	4-ClPh	2	CH ₃ CN, P ₂ O ₅	3	6d	18
1a	Ph	1	DMF	1	7a	30
1b	SMe	1	DMF	1	7 b	21
1c	4- MePh	1	DMF	1	7c	24
1d	4-ClPh	1	DMF	1	7 d	22

oxidative agents as copper(II) acetate, nitrobenzene, or air oxygen have no effect on the reaction. ¹⁸ We demonstrated the

same to be true for DDQ and CAN $((NH_4)_2Ce(NO_3)_6)$. It turned out that if protic acids in the reaction are replaced by Lewis acids, such as AlCl₃, the same product **6a** is formed. Our results indicate that the most efficient route to cycloaddition products **6** and **7** in zwitterionic form is performing the reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a-d) with ketones **4** and **5** in the presence of *p*-TsOH either in refluxing DMF or in refluxing acetonitrile (Tables 1 and 2). The addition of P_2O_5 in the reaction mixture as a dehydrating agent leads to increase in the yield of **6a,c,d**.

If strong acids were used in the reaction, salts of 6a-d were formed. There was no noticed influence of acid strength on the reaction yield. The reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a-d) with ketone 4 in the presence of trifluoromethane-sulfonic acid leads to the salts 6aa-da (Table 3). Reducing the

Table 3. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a-e) with Cyclohexanone (4) in the Presence of 1 equiv of CF_3SO_3H ($X^- = CF_3SO_3^-$)

starting material	R	time, days	product	yield, %
1a	Ph	7	6aa	8
1a	Ph	17	6aa	18
1a	Ph	35	6aa	26
1a	Ph	35 ^a	6aa	21
1a	Ph	35 ^b	6aa	3
1a	Ph	35 ^c	6aa	25
1b	SMe	35	6ba	24
1c	4-MePh	35	6ca	26
1d	4-ClPh	35	6da	26
1e	CH_2Ph	35	-	0

 a 3 equiv of triflic acid. b 1 equiv of cyclohexanone. c 4 equiv of cyclohexanone.

amount of cyclohexanone in the reaction mixture from 2 equiv to 1 equiv leads to a significant decrease of **6aa** yield. Increasing cyclohexanone or CF_3SO_3H concentrations does not have visible effects. Compounds **6ab**—**db** are soluble in acidic solutions, and neutralization of reaction mixture leads to an increased yield of products **6a**—**d**. Triazines with aliphatic substituents in position 3 (**1e**, R = CH_2Ph) do not react with cyclic ketones either in the presence of CF_3SO_3H or in ethanolic HCl solution. 1,2,4-Triazines without an oxo group in position 5 of the triazine ring (3-Ph-1,2,4-triazine, 3-SMe-1,2,4-triazine) also have no reactivity under such conditions.

It is known cyclohexanone 4 gives condensation product 9 in the presence of acids, for example, in the presence of trifluoromethanesulfonic acid (Scheme 2). It was logical to

Scheme 2

expect the formation of products $6\mathbf{a}$ — \mathbf{d} in the reaction of 3-R-1,2,4-triazin-5(4H)-one (1) with 1,1'-bi(cyclohexylidene)-2-one (9), and we obtained them in refluxing DMF with addition of CF₃SO₃H (15% yield), but at ambient temperature in the presence of CF₃COOH the reaction leads to lactams $10\mathbf{a}$ — \mathbf{c} , \mathbf{e} (Table 4).

Table 4. Reaction of 3-R-1,2,4-triazin-5(4H)-ones (1a-c,e) with 9a in the Presence of CF₃COOH

starting material	R	product	yield, %
1a	Ph	10a	58
1b	SMe	10b	71
1c	4-MePh	10c	68
1e	CH_2Ph	10e	53

Our next step was to study the reaction of 1'-hydroxybi-(cyclohexan)-2-one (8) with 3-Ph-1,2,4-triazin-5(4H)-one (1a). When the reaction was carried out at ambient temperature in the presence of CF₃COOH/CF₃SO₃H, the formation of 6a was not observed. Reaction in refluxing CH₃CN in the presence of P₂O₅ resulted in the formation of 6a (54% yield) (Scheme 3).

Scheme 3

When we tried to use 1,1'-bi(cyclopentylidene)-2-one (11) instead of 1,1'- bi(cyclohexylidene)-2-one (9) in the reaction with 3-R-1,2,4-triazin-5(4H)-ones, only product 12c (R = 4-MePh) could be isolated and identified. In all other cases we could not obtain pure products (Scheme 4). The structure of

Scheme 4

compound 12c was assigned by analogy with compounds 10a–c and confirmed by mass spectrometry data and ¹H and ¹³C NMR spectra. When we attempted to carry out the reaction under other conditions, we observed no reaction in refluxing acetonitrile in the presence of CF₃COOH. The reaction was carried out in refluxing DMF in the presence of CF₃SO₃H and resulted in the formation of 7a triflate (19% yield) after 1 h.

To prove the assumption that compounds 10 are intermediate compounds in the reactions leading to 6 we investigated their behavior in different conditions. It appears that 10a does not undergo transformations in refluxing DMF or in DMF at ambient temperature in the presence of CF₃COOH or *p*-TsOH. Refluxing of 10a in DMF in the presence of CF₃SO₃H results in the formation of triflate of triazinone 1a and the condensation product 6aa in low yields (Scheme 5).

Scheme 5

The salt of 1a is a decomposition product of 10a and could be used as starting material for reaction with cyclohexanone residue from the reaction mixture resulting in the formation of 632

Liquid chromatography-mass spectrometry analysis of the reaction mixtures of 3-Ph-1,2,4-triazin-5(4H)-one (1a) with cyclohexanone in DMSO solutions in the presence of CF₃SO₃H after 24 h shows intensive peaks of some molecular ions. There were observed peaks of 6ab (ESI-MS, m/z =332.1798 (calcd 332.1757 for $[C_{21}H_{22}N_3O]^+$, $[M + H]^+$), 10a (ESI-MS, m/z = 352.2032 (calcd 352.2020 for $[C_{21}H_{26}N_3O_2]^+$, $[M + H]^{+}$), and unidentified compound (ESI-MS, m/z =258.1610 (calcd 258.1601 for $[C_{15}H_{20}N_3O]^+$, $[M + H]^+$)). In the reaction of 3-Ph-1,2,4-triazin-5(4H)-one (1a) with cyclopentanone, intensive peaks of 7ab were observed (ESI-MS, m/z= 304.1488 (calcd 304.1444 for $[C_{19}H_{18}N_3O]^+$, $[M + H]^+$)), 12a (ESI-MS, m/z = 324.1729 (calcd 324.1706 for $[C_{19}H_{22}N_3O_2]^+$, $[M + H]^+$), benzonitrile (ESI-MS, m/z =104.0520 (calcd 104.0494 for $[C_7H_6N]^+$, $[M+H]^+$)), peak of unidentified compound (ESI-MS, m/z = 244.1470 (calcd 244.1444 for $[C_{14}H_{18}N_3O]^+$, $[M + H]^+$), its dimer (ESI-MS, m/z = 477.2034 (calcd 477.2033 for $[C_{28}H_{25}N_6O_2]^+$, [M +H]⁺)), and peak of unidentified compound (ESI-MS, m/z =545.2625 (calcd 545.2619 for $[C_{28}H_{33}N_8O_4]^+$, $[M + H]^+$).

Absence of the formation of lactam products 10 and 12 in quantities sufficient for identification in the reactions of triazines 1 with cyclic ketones probably means that there are two different ways of formation of tetracyclic compounds 6 and 7 and lactam compounds 10 and 12. We suggested two different routes for the reactions of triazines 1 with free cyclic ketones and their dimerization products.

The high activity of 3-R-1,2,4-triazin-5(4H)-ones in the reactions with C-nucleophiles suppose initial nucleophilic addition of cyclic ketones to 1,2,4-triazine ring with formation of high reactive compounds **1A** or **1B**. The attack of the cyclic ketone on **1A** leads to the series of transformations with the final product **6**. We suppose that the product **1B** after intermolecular addition gives **1D**. Compound **1D** at ambient temperature gives 10-membered lactame **10**; the high temperature of refluxing DMF and the presence of triflic acid lead to the loss of a water molecule and formation of **6** (Scheme 6).

CONCLUSIONS

In the course of the present study, new heterocyclic systems of benzo[c][1,2,4]triazino[1,6-a][2]azecine and cyclopenta[c]-

Scheme 6

[1,2,4]triazino[1,6-a]azonine were obtained, optimal reaction conditions were found for the condensation between 1,2,4-triazine-5(4H)-ones and cyclic ketones. It was shown that benzo[c][1,2,4]triazino[1,6-a][2]azecines 10 are not intermediates in the reaction of formation of [1,2,4]triazino[1,6-f]phenanthridine derivatives 6.

It should be noted that C–C coupling reactions between azines and cyclic ketones have been described previously,²⁰ but intramolecular condensation was observed and investigated for the first time.

EXPERIMENTAL SECTION

3-Ph-1,2,4-triazin-5(4H)-one (1a), ²¹ 3-SMe-1,2,4-triazin-5(4H)-one (1b), ²² 1'-hydroxybi(cyclohexan)-2-one (8), ²³ 1,1'-bi-(cyclohexylidene)-2-one (9), ²⁴ and 1,1'-bi(cyclopentilidene)-2-one (11)²⁵ were synthesized by known methods; other starting materials are commercially available. ¹H and ¹³C NMR spectra were recorded using 400 MHz spectrometer; tetramethylsilane (TMS) was used as an internal standard. A TOF mass analyzer was used for the HRMS.

3-(4-Tolyl)-1,2,4-triazin-5(4H)-one (1c). Methanol (10.1 mL, 0.170 mol) was added to the solution of 4-methylbenzonitrile (10.0 g, 0.085 mol) in ether (50 mL), the resulting solution was chilled to 0 to -5 °C, and then dry HCl was bubbled through the solution for 2–3 h. The reaction mixture was stirred for 16 h at room temperature. The sediment of methyl ether of 4-methylbenzimidate hydrochloride was filtered out, washed with ether, and dried. The hydrochloride was dissolved in water, and 50 mL of aqueous NaOH (5 N) was added to the solution. The product was extracted with ether, the ether solution was dried over Na₂SO₄, and ether was evaporated.

At the next stage, hydrazine hydrate (2.35 g, 0.047 mol) was added to solution of 4-methylbenzimidate (7.0 g, 0.047 mol) in 10 mL of methanol. The reaction mixture was stirred for 16 h it was then cooled to 5 °C, and glyoxylic acid monohydrate (4.32 g, 0.047 mmol) was added to the solution in portions. The temperature has to be lower 10 °C. The reaction mixture was stirred for 1 h at 5 °C, then it was stored in refrigerator for 16 h at 5 °C. The formed yellow sediment was filtered, washed with cold methanol, and dried. The yellow solid was dissolved in DMF (50 mL) and refluxed for 30 min. After cooling of the solution was cooled to room temperature crystals formed. They were filtered and crystallized from ethanol: yield 41% (6.5 g); light brown crystals; mp 255 °C; 1 H NMR (400 MHz, DMSO- d_6) δ 2.43 (s,

3H), 7.35 (d, J = 8.1 Hz, 2H), 7.63 (s, 1H), 7.98 (d, J = 8.2 Hz, 2H), 14.0 (br.s, 1H, NH); 13 C NMR (100 MHz, DMSO- d_6) δ 21.0, 127.5, 127.6, 129.5, 143.0, 143.4, 158.1, 162.2; ESI-MS m/z = 188.0819 (calcd 188.0818 for [C_{10} H $_{10}$ N $_{3}$ O] $^{+}$, [M + H] $^{+}$).

3-(4-Chlorophenyl)-1,2,4-triazin-5(4*H***)-one (1d).** The procedure is the same as for 1c: yield 31% (5.4 g); yellow crystals; mp 282–283 °C; ¹H NMR (400 MHz, DMSO- d_6) δ 7.68 (d, 2H), 7.86 (s, 1H), 8.07 (d, 2H), 14.12 (br.s, 1H); ¹³C NMR (100 MHz, DMSO- d_6) δ 129.6, 129.8, 130.1, 138.1, 144.3, 157.8, 162.5; ESI-MS m/z = 208.0283 (calcd 208.0272 for $[C_9H_7ClN_3O]^+$, $[M + H]^+$).

3-Benzyl-1,2,4-triazin-5(4H)-one (1e). The procedure is the same as for 1c: yield 39% (6.2 g); yellow crystals; mp 165–166 °C; 1 H NMR (400 MHz, DMSO- d_6) δ 3.11 (s, 2H), 6.43–6.54 (m, 5H), 6.92 (s, 1H), 13.68 (br.s, 1H); 13 C NMR (100 MHz, CD₃OD) δ 41.0, 128.7, 130.0, 130.1, 135.9, 144.7, 165.3, 165.4; ESI-MS m/z = 188.0812 (calcd 188.0818 for $[C_{10}H_{10}N_3O]^+$, $[M+H]^+$).

6-(2-Oxo-2-phenylethyl)-3-phenyl-1,6-dihydro-1,2,4-triazin-5(4H)-one (3a). Acetophenone (270 μ L, 2.310 mmol) and CF₃SO₃H (100 μ L) were added to the suspension of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in 10 mL of CH₃CN. The solution was refluxed for 3 h, then it was cooled and neutralized with NEt₃, and after 16 h the formed sediment was filtered and crystallized from CH₃CN: yield 35% (119 mg); yellow crystals; mp 194–195 °C; ¹H NMR (400 MHz, CDCl₃) δ 3.30 (dd, J = 18.5, 10.1 Hz, 1H), 3.92 (dd, J = 18.5, 2.4 Hz, 1H), 4.29 (dd, J = 10.1, 2.3 Hz, 1H), 6.39 (br.s, 1H), 7.41–7.45 (m, 3H), 7.48–7.52 (m, 2H), 7.60–7.64 (m, 1H), 7.67–7.70 (m, 2H), 8.00–8.02 (m, 2H), 8.37 (br.s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 37.1, 52.3, 124.9, 128.2, 128.8, 129.0, 130.1, 131.2, 133.9, 136.0, 139.4, 167.7, 197.6; ESI-MS m/z = 294.1291 (calcd 294.1237 for $[C_{17}H_{16}N_3O_2]^+$, $[M + H]^+$).

3-(Methylthio)-6-(2-oxo-2-phenylethyl)-1,6-dihydro-1,2,4-triazin-5(4H)-one (3b). Acetophenone (326 μ L, 2.794 mmol) and CF₃SO₃H (100 μ L) were added to the suspension of 3-SMe-1,2,4-triazin-5(4H)-one (200 mg, 1.397 mmol) in 10 mL of CH₃CN. The solution was refluxed for 3 h (emission of CH₃SH !), and then it was cooled, neutralized with NEt₃, and evaporated. The residue was chromatographed by column using EtOAc as an eluent: yield 13% (40 mg); yellow crystals; mp 132–133 °C; R_f = 0.3 (EtOAc); ¹H NMR (400 MHz, DMSO- d_6) δ 2.33 (s, 3H), 3.10 (dd, J = 17.8, 6.0 Hz, 1H), 3.57 (dd, J = 17.8, 5.8 Hz, 1H), 3.86–3.89 (m, 1H), 6.56 (d, J = 3.0 Hz, 1H), 7.48–7.52 (m, 2H), 7.58–7.62 (m, 1H), 7.98–7.99 (m, 2H), 10.64 (s, 1H); ¹³C NMR (100 MHz, DMSO- d_6) δ 12.8, 36.4, 52.2, 128.0, 128.7, 133.3, 136.3, 139.3, 167.3, 196.6; ESI-MS m/z = 264.0791 (calcd 264.0801 for [C₁₂H₁₄N₃O₂S]⁺, [M + H]⁺).

Procedures for Synthesis of 6a. Method A. Cyclohexanone 4 (239 μ L, 2.310 mmol) and 1 equiv of acid (CF₃COOH or AlCl₃, see Table 1) were added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in the corresponding solvent (see Table 1). The solution was stirred or refluxed for the time mentioned in the Table 1. Formed sediment was filtered and crystallized from DMF. Otherwise, the reaction mixture was poured to the water (50 mL) and extracted with CH₂Cl₂. The organic layer was separated, washed with water and brine, and dried over Na₂SO₄. The solution in CH₂Cl₂ was evaporated and the residue was crystallized from DMF.

 \dot{M} ethod B. Cyclohexanone 4 (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4 \dot{H})-one (200 mg, 1.155 mmol) in phosphoric acid (5 mL). The solution was stirred for 3 h at 50 °C, and then it was cooled and poured into water (50 mL). Formed sediment was filtered and column chromatographed using EtOAc as an eluent.

Method C. Cyclohexanone 4 (239 μL, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in acetic acid (5 mL). The reaction mixture was refluxed for 1 h, and then it was evaporated. The residue was refluxed in CHCl₃ for 5 min and filtered. The chloroform solution was evaporated, and the residue was crystallized from DMF.

Method D. 1,1'-Bi(cyclohexylidene)-2-one (9) (206 μ L, 1.155 mmol) and CF₃SO₃H (1.155 mmol, 102 μ L) were added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in DMF (3 mL). The reaction mixture was refluxed for 1 h, neutralized with NEt₃, and evaporated. The residue was crystallized from DMF.

Method E. 1'-Hydroxybi(cyclohexan)-2-one (8) (227 μL, 1.155 mmol), p-TsOH (203 mg, 1.155 mmol), and P_2O_5 (492 mg, 1.155 mmol) were added to the mixture of 3-R-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) and CH₃CN (10 mL). It was then divided into two parts. The solution in CH₃CN was decanted and evaporated to half volume. NEt₃ (100 μL) was added to the solution, and after 2 h the sediment was filtered. The solid part of the reaction mixture was dissolved in water and neutralized with a concentrated solution of NaHCO₃. The resulting sediment was filtered. Combined sediments were dried and crystallized from DMF.

Method F. Cyclohexanone 4 (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4H)-one (200 mg, 1.155 mmol) in methanolic MeONa (1M) (4 mL). The reaction mixture was stirred for 24 h, poured into water (50 mL), and neutralized with HCl (2 M) and the product extracted with CH₂Cl₂. The solution in CH₂Cl₂ was washed with water and brine, dried over Na₂SO₄, and then evaporated, and the residue was crystallized from DMF.

Procedures for Synthesis of 6a–d and 7a–d. Method A. Ketone 4 or 5 (2 equiv), p-TsOH (1 equiv), and P_2O_5 (3 equiv) (or 1 g of molecular sieves 3 Å) were added to the mixture of 3-R-1,2,4-triazin-5(4H)-one (150 mg) and CH₃CN (10 mL). The reaction mixture was refluxed for 3 h. it was then divided into two parts. The solution in CH₃CN was decanted and evaporated to half its volume. NEt₃ (100 μ L) was added to the solution, and after 2 h the sediment was filtered. The solid part of the reaction mixture was dissolved in water and neutralized with a concentrated solution of NaHCO₃; the resulting sediment was filtered. The combined sediments were dried and crystallized from DMF.

Method B. Ketone 4 or 5 (2 equiv) and p-TsOH (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 mL). The solution was refluxed for 1 h. After 16 h, sediment was filtered. If there was no sediment the reaction mixture was poured into water, and the product extracted with CH_2Cl_2 . The organic layer was separated, washed with water and brine, and dried over Na_2SO_4 . The solution in CH_2Cl_2 was evaporated, and the residue was crystallized from DMF.

2-Phenyl-5,6,7,8,9,10,11,12-octahydro[1,2,4]triazino[1,6-*f*]-**phenanthridin-13-ium-4-olate** (6a): yield 44% (169 mg); cream crystals; mp >300 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.77–1.81 (m, 4H), 1.90–1.99 (m, 4H), 2.67 (t, J = 6.1 Hz, 2H), 2.71 (t, J = 6.2 Hz, 2H), 3.35 (t, J = 6.2 Hz, 2H), 3.61 (t, J = 6.2 Hz, 2H), 7.38–7.46 (m, 3H), 8.36–8.38 (m, 2H). The spectral data of the compound are identical to literature values. ¹⁷

2-(Methylthio)-5,6,7,8,9,10,11,12-octahydro[1,2,4]triazino-[**1,6-f]phenanthridin-13-ium-4-olate (6b):** yield 21% (73 mg); cream crystals; mp 249 °C; 1 H NMR (400 MHz, CDCl₃) δ 1.76–1.97 (m, 8H), 2.54 (s, 3H), 2.74 (dd, J = 13.3, 6.8 Hz, 4H), 3.19 (t, J = 6.2 Hz, 2H), 3.59 (t, J = 6.2 Hz, 2H); 13 C NMR (100 MHz, DMSO- d_6) δ 13.5, 21.0, 21.2, 21.4, 21.9, 26.3, 27.1, 27.7, 28.8, 131.4, 136.1, 136.4, 142.4, 146.6, 163.8, 170.9; ESI-MS m/z = 302.1355 (calcd 302.1322 for $[C_{16}H_{20}N_3OS]^+$, $[M+H]^+$).

2-p-Tolyl-5,6,7,8,9,10,11,12-octahydro[1,2,4]triazino[1,6-f]-**phenanthridin-13-ium-4-olate** (6c): yield 24% (96 mg); gray crystals; mp 304–305 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.77–1.79 (m, 4H), 1.90–1.99 (m, 4H), 2.38 (s, 3H), 2.72–2.78 (m, 4H), 3.39 (t, J = 6.3 Hz, 2H), 3.66 (t, J = 6.1 Hz, 2H), 7.19 (d, J = 8.0 Hz, 2H), 8.22 (d, J = 8.2 Hz, 2H); ¹³C NMR (100 MHz, DMSO- d_6) δ 21.0, 21.3, 21.5, 21.9, 26.3, 27.1, 27.6, 28.8, 128.0, 128.7, 132.1, 132.9, 135.7, 136.3, 141.1, 143.3, 146.6, 161.5, 166.5; ESI-MS m/z = 346.1903 (calcd 346.1914 for $[C_{12}H_{24}N_3O]^+$, $[M+H]^+$). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition number CCDC 859049). These data can be obtained via www.ccdc.cam.ac.uk/data request/cif.

2-(4-Chlorophenyl)-5,6,7,8,9,10,11,12-octahydro[1,2,4]-triazino[1,6-f]phenanthridin-13-ium-4-olate (6d): yield 15% (63 mg); yellow crystals; mp 257–258 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.76–1.96 (m, 8H), 2.76–2.77 (m, 4H), 3.30 (t, J = 5.7 Hz, 2H), 3.45 (t, J = 5.9 Hz, 2H), 7.44 (d, J = 8.5 Hz, 2H), 8.23 (d, J = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 21.0, 21.3, 21.5, 21.9, 26.4, 27.1, 27.7, 28.9, 128.0, 129.2, 132.0, 134.1, 135.9, 136.7, 136.9, 143.6, 147.5,

160.2, 166.5; ESI-MS m/z = 366.1354 (calcd 366.1368 for $[C_{21}H_{21}ClN_3O]^+$, $[M + H]^+$).

2-Phenyl-5,6,7,8,9,10-hexahydrodicyclopenta[3,4:5,6]-pyrido[2,1-f][1,2,4]triazin-11-ium-4-olate (7a): yield 30% (105 mg); light green crystals; mp >300 °C; 1 H NMR (400 MHz, CDCl₃) δ 2.25–2.33 (m, 2H), 2.37–2.45 (m, 2H), 3.01 (t, J=7.7 Hz, 2H), 3.09–3.18 (m, 2H), 3.55 (t, J=7.9 Hz, 2H), 3.79 (t, J=7.9 Hz, 2H), 7.40–7.47 (m, 3H), 8.42–8.44 (m, 2H). The spectral data of the compound are identical to literature values. 17

2-(Methylthio)-5,6,7,8,9,10-hexahydrodicyclopenta[3,4:5,6]-pyrido[2,1-f][1,2,4]triazin-11-ium-4-olate (7b): yield 21% (66 mg); gray crystals; mp 258–259 °C; ¹H NMR (400 MHz, CDCl₃) δ 2.21–2.30 (m, 2H), 2.31–2.42 (m, 2H), 2.53 (s, 3H), 2.98 (t, J = 7.8 Hz, 2H), 3.10 (t, J = 7.6 Hz, 2H), 3.37 (t, J = 7.8 Hz, 2H), 3.70 (t, J = 7.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 13.5, 22.1, 24.7, 30.5, 31.2, 31.5, 34.0, 129.4, 139.2, 143.2, 149.4, 152.5, 162.8, 172.3. The spectral data of the compound are identical to literature values.¹⁷

2-*p*-Tolyl-5,6,7,8,9,10-hexahydrodicyclopenta[3,4:5,6]-pyrido[2,1-f][1,2,4]triazin-11-ium-4-olate (7c): yield 24% (88 mg); brown crystals; mp 236–237 °C; 1 H NMR (400 MHz, CDCl₃) δ 2.20–2.27 (m, 2H), 2.34–2.42 (m, 5H), 2.88–2.96 (m, 2H), 3.08 (t, J = 7.6 Hz, 2H), 3.51 (t, J = 7.7 Hz, 2H), 3.74 (t, J = 7.6 Hz, 2H), 7.23 (d, J = 8.0 Hz, 2H), 8.28 (d, J = 8.1 Hz, 2H); 13 C NMR (100 MHz, CDCl₃) δ 21.4, 22.1, 24.5, 30.4, 31.3, 31.3, 34.1, 128.0, 128.7, 130.2, 133.0, 139.0, 141.1, 142.8, 150.4, 152.5, 162.8, 165.6; ESI-MS m/z = 318.1620 (calcd 318.1601 for [C_{20} H₂₀N₃O] $^{+}$, [M + H] $^{+}$).

2-(4-Chlorophenyl)-5,6,7,8,9,10-hexahydrodicyclopenta-[**3,4:5,6]pyrido**[**2,1-f**][**1,2,4]triazin-11-ium-4-olate** (**7d):** yield 22% (86 mg); yellow green crystals; mp >300 °C; ¹H NMR (400 MHz, CDCl₃) δ 2.21–2.28 (m, 2H), 2.34–2.41 (m, 2H), 2.94 (t, J = 7.7 Hz, 2H), 3.07 (t, J = 7.6 Hz, 2H), 3.47 (t, J = 7.7 Hz, 2H), 3.70 (t, J = 7.6 Hz, 2H), 7.34 (d, J = 8.6 Hz, 2H), 8.26 (d, J = 8.6 Hz, 2H). The spectral data of the compound are identical to literature values. ¹⁷

Procedures for Synthesis of 6aa–da. *Method A.* Cyclohexanone (2 equiv) and CF_3SO_3H (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 mL). The reaction mixture was stirred at room temperature for the period of time specified in Table 3. The sediment was filtered and crystallized from DMF.

Method B. Cyclohexanone (2 equiv) and CF_3SO_3H (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 mL). The reaction mixture was refluxed for the time in Table 1 and cooled; after 12 h, the formed sediment was filtered and crystallized from DMF.

4-Oxo-2-phenyl-3,4,5,6,7,8,9,10,11,12-decahydro[1,2,4]-triazino[1,6-f]phenanthridin-13-ium trifluoromethanesulfonate (6aa): yield 25% (138 mg); green crystals; mp 252–253 °C; 1 H NMR (400 MHz, CDCl₃) δ 1.78–2.04 (m, 8H), 2.85–2.66 (m, 4H), 3.40 (t, J = 6.4 Hz, 2H), 3.67 (t, J = 6.1 Hz, 2H), 7.42–7.50 (m, 3H), 8.41–8.44 (m, 2H); 13 C NMR (100 MHz, CDCl₃) δ 21.1, 21.3, 21.6, 22.0, 26.3, 27.2, 27.6, 28.8, 128.0, 128.1, 130.8, 135.9, 136.0, 136.3, 143.4, 146.9, 161.7, 166.4; 19 F NMR (100 MHz, CDCl₃) δ –78.2; ESI-MS m/z = 332.1762 (calcd 332.1757 for [C₂₁H₂₂N₃O]⁺, [M + H]⁺). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition no. CCDC 820437). These data can be obtained via www.ccdc.cam.ac.uk/data_request/cif.

2-(Methylthio)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[**1,2,4]triazino**[**1,6-f]phenanthridin-13-ium trifluoromethane-sulfonate (6ba):** yield 24% (150 mg); light yellow crystals; mp 215–216 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.74–1.80 (m, 2H), 1.84–1.95 (m, 6H), 2.52 (s, 3H), 2.72–2.77 (m, 4H), 3.16–3.19 (m, 2H), 3.54–3.57 (m, 2H); 13 C NMR (100 MHz, CDCl₃) δ 13.6, 21.1, 21.3, 21.4, 22.0, 26.3, 27.2, 27.7, 28.9, 131.4, 136.2, 136.6, 146.2, 146.8, 163.8, 170.9; 19 F NMR (100 MHz, CDCl₃) δ –78.3; ESI-MS m/z=302.1318 (calcd 302.1322 for $[\mathrm{C_{16}H_{20}N_3OS}]^+$, $[\mathrm{M}+\mathrm{H}]^+$).

2-(4-Methylphenyl)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[**1,2,4]triazino**[**1,6-**f]**phenanthridin-13-ium trifluoromethane-sulfonate (6ca):** yield 26% (140 mg); light yellow crystals; mp 277–278 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.72–1.83 (m, 4H), 1.87–2.01 (m, 4H), 2.38 (s, 3H), 2.62–2.65 (m, 2H), 2.68–2.71 (m,

2H), 3.31–3.34 (m, 2H), 3.57–3.60 (m, 2H), 7.19 (d, J = 7.6 Hz, 2H), 8.22 (d, J = 8.0 Hz, 2H); NMR ¹³C (100 MHz, CDCl₃) δ 21.0, 21.3, 21.5, 22.0, 26.3, 27.2, 27.7, 28.9, 128.0, 128.8, 132.1, 132.9, 135.8, 136.4, 141.2, 143.5, 146.8, 161.4, 166.5; ¹⁹F NMR (100 MHz, CDCl₃) δ –78.3; ESI-MS m/z = 346.1914 (calcd 346.1914 for [C₂₂H₂₄N₃O]⁺, [M + H]⁺).

2-(4-Chlorophenyl)-4-oxo-3,4,5,6,7,8,9,10,11,12-decahydro-[**1,2,4**]triazino[**1,6-**f]phenanthridin-**13-ium** trifluoromethane-sulfonate (**6da**): yield 26% (131 mg); wine red crystals; mp 252–253 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.75–1.98 (m, 8H), 2.65–2.71 (m, 4H), 3.25 (t, J = 6.0 Hz, 2H), 3.54–3.57 (m, 2H), 7.30 (d, J = 8.4 Hz, 2H), 8.22 (d, J = 8.8 Hz, 2H); 13 C NMR (100 MHz, CDCl₃) δ 21.0, 21.3, 21.5, 21.9, 26.4, 27.1, 27.7, 28.9, 128.0, 129.2, 132.0, 134.1, 135.9, 136.7, 136.9, 143.6, 147.5, 160.2, 166.5; 19 F NMR (100 MHz, CDCl₃) δ -77.3; ESI-MS m/z = 366.1372 (calcd 366.1368 for $[C_{21}H_{21}ClN_3O]^+$, $[M+H]^+$).

Procedure for Synthesis of 4-Oxo-2-phenyl-3,4,5,6,7,8,9,10,11,12-decahydro[1,2,4]triazino[1,6-f]-phenanthridin-13-ium Chloride (6ab). Cyclohexanone (239 μ L, 2.310 mmol) was added to the solution of 3-Ph-1,2,4-triazin-5(4*H*)-one (200 mg, 1.155 mmol) in concentrated ethanol solution of HCl (5 mL). The reaction mixture was stirred at room temperature for 1 day. The sediment was filtered, washed with ethanol, crystallized from DMF, and dried: yield 40% (172 mg); dark yellow crystals; mp 270–271 °C; ¹H NMR (400 MHz, DMSO- d_6) δ 1.86–2.01 (m, 8H), 2.92–2.98 (m, 4H), 3.42–3.45 (m, 2H), 3.49–3.51 (m, 2H), 7.59–7.63 (m, 2H), 7.69–7.73 (m, 1H), 8.17 (d, J = 7.2 Hz, 2H); 13 C NMR (100 MHz, DMSO- d_6) δ 20.6, 20.8, 21.1, 21.5, 26.4, 27.8, 28.0, 28.5, 128.7, 129.1, 129.1, 129.6, 132.7, 133.8, 137.2, 140.7, 147.7, 152.7, 154.1, 157.4, 157.5; ESI-MS m/z = 332.1769 (calcd 332.1757 for $[C_{21}H_{22}N_3O]^+$, $[M+H]^+$).

Procedure for Synthesis of 10a–c,e. 1,1'-Bi(cyclohexilidene)-2-one (9) (1 equiv) and CF₃COOH (1 equiv) were added to the solution of 3-R-1,2,4-triazin-5(4H)-one (200 mg) in DMF (3 mL). The reaction mixture was stirred at room temperature for 5 days, poured into a Petri dish; residue from the flask was washed off with ethanol (5 mL) into the same Petri dish. After 2 h, the formed sediment was filtered and crystallized from CH₃CN.

3-Phenyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1*H***-benzo[c][1,2,4]triazino[1,6-a]azecine-1,6(2H)-dione (10a):** yield 57% (232 mg); colorless crystals; mp 210–211 °C; ¹H NMR (400 MHz, CDCl₃) δ 1.15–1.27 (m, 1H), 1.41–2.14 (m, 14H), 2.28–2.32 (m, 1H), 3.56 (dd, J = 14.3, 10.1, 1.5 Hz, 1H), 4.90 (d, J = 10.1 Hz, 1H), 5.12 (d, J = 11.4 Hz, 1H), 7.41–7.47 (m, 3H), 7.88–7.90 (m, 2H), 11.30 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 21.6, 24.7, 25.5, 25.8, 26.3, 27.6, 29.0, 32.4, 43.1, 57.0, 125.9, 129.0, 130.3, 130.4, 131.0, 135.1, 139.2, 167.8, 176.0; ESI-MS m/z = 352.1996 (calcd 352.2020 for $[C_{21}H_{26}N_3O_2]^+$, $[M+H]^+$). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition no. CCDC 859113). These data can be obtained via www.ccdc.cam.ac.uk/data request/cif.

3-(Methylthio)-7,8,9,10,12,13,14,15,15a,15b-decahydro-1*H***-benzo[c][1,2,4]triazino[1,6-***a*]**azecine-1,6(2***H*)**-dione (10b):** yield 70% (315 mg); colorless crystals; mp 244–245 °C; ¹H NMR (400 MHz, DMSO- d_6) δ 1.11–1.21 (m, 1H), 1.33–1.41 (m, 1H), 1.46–2.12 (m, 13H), 2.28 (d, J = 15.0 Hz, 1H), 2.46 (s, 3H), 3.33 (dd, J = 14.0, 9.6 Hz, 1H), 4.90 (d, J = 10.4 Hz, 1H), 5.01 (d, J = 11.4 Hz, 1H), 11.30 (br.s, 1H); ¹³C NMR (100 MHz, DMSO- d_6) δ 13.6, 21.6, 24.5, 25.5, 25.7, 26.0, 27.4, 28.9, 32.2, 42.8, 57.5, 129.9, 135.2, 141.3, 166.1, 175.0; ESI-MS m/z = 322.1580 (calcd 322.1584 for $[C_{16}H_{24}N_3O_2S]^+$, $[M+H]^+$). Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (deposition no. CCDC 820598). These data can be obtained via www.ccdc.cam.ac.uk/data_request/cif.

3-p-Tolyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1H-benzo[c][1,2,4]triazino[1,6-a]azecine-1,6(2H)-dione (10c): yield 67% (262 mg); colorless crystals; mp 205–206 °C; [α]_D = 0 (c = 0.6, MeOH); 1 H NMR (400 MHz, DMSO- d_6) δ 1.11–1.28 (m, 1H), 1.39–2.15 (m, 14H), 2.30 (d, J = 14.5 Hz, 1H), 2.40 (s, 3H), 3.55 (dd, J = 14.1, 10.0 Hz, 1H), 4.89 (d, J = 10.3 Hz, 1H), 5.11 (d, J = 11.4 Hz,

1H), 7.22 (d, J = 8.1 Hz, 2H), 7.77 (d, J = 8.1 Hz, 2H), 11.24 (s, 1H); ¹³C NMR (100 MHz, DMSO- d_6) δ : 20.6, 21.1, 23.9, 24.9, 25.1, 25.4, 26.9, 28.3, 31.5, 41.8, 56.3, 126.2, 127.5, 129.1, 129.2, 134.4, 140.4, 140.5, 165.9, 174.7; ESI-MS, m/z = 366.2143 (calcd 366.2176 for $[C_{22}H_{28}N_3O_7]^+$, $[M + H]^+$).

3-Benzyl-7,8,9,10,12,13,14,15,15a,15b-decahydro-1*H***-benzo[c][1,2,4]triazino[1,6-***a*]**azecine-1,6(2***H*)**-dione (10e):** yield 52% (203 mg); light brown crystals; mp 211–212 °C; ¹H NMR (400 MHz, DMSO- d_6) δ 1.03–1.15 (m, 1H), 1.27–2.09 (m, 14H), 2.23–2.27 (m, 1H), 3.61 (s, 2H), 4.85 (d, J = 10.1 Hz, 1H), 4.96 (d, J = 11.4 Hz, 1H), 7.24–7.29 (m, 1H), 7.34–7.35 (m, 4H), 11.1 (s, 1H); ¹³C NMR (100 MHz, DMSO- d_6) δ 21.0, 23.9, 24.8, 25.0, 25.2, 26.9, 28.2, 31.3, 37.5, 41.7, 55.8, 126.8, 128.3, 128.7, 129.1, 134.3, 135.8, 142.9, 165.6, 174.4; ESI-MS m/z = 366.2142 (calcd 366.2176 for $[C_{22}H_{28}N_3O_2]^+$, $[M+H]^+$).

3-p-Tolyl-7,8,9,11,12,13,13a,13b-octahydro-1*H*-cyclopenta-[c][1,2,4]triazino[1,6-a]azonine-1,6(2H)-dione (12c). 1,1'-Bi-(cyclopentylidene)-2-one (11) (162 μ L, 1.068 mmol) and CF₃COOH (100 μ L) were added to the solution of 3-(4-Tol)-1,2,4-triazin-5(4H)one (1c) (200 mg, 1.068 mmol) in DMF (3 mL). The reaction mixture was stirred at room temperature for 5 days and evaporated. The residue was chromatographed by column using EtOAc as an eluent: yield 9% (32 mg); yellow crystals; mp 229–230 °C; $R_f = 0.6$ (EtOAc); ¹H NMR (400 MHz, DMSO- d_6) δ 1.64–1.75 (m, 4H), 1.77-1.85 (m, 1H), 2.08-2.15 (m, 1H), 2.23-2.44 (m, 7H), 2.58-2.64 (m, 1H), 2.71 (br.s, 2H), 3.71 (dd, *J* = 5.5, 2.1 Hz, 1H), 6.99 (d, *J* = 2.1 Hz, 1H), 7.14 (d, *J* = 8.1 Hz, 2H), 7.61 (d, *J* = 8.2 Hz, 2H), 10.56 (s, 1H); 13 C NMR (100 MHz, DMSO- d_6) δ 20.5, 22.1, 24.4, 26.1, 26.7, 31.7, 33.3, 47.4, 51.7, 56.5, 99.2, 124.9, 127.3, 128.5, 129.2, 138.3, 138.9, 157.3, 165.7, 203.7; ESI-MS m/z = 338.1849 (calcd 338.1863 for $[C_{20}H_{24}N_3O_2]^+$, $[M + H]^+$).

ASSOCIATED CONTENT

S Supporting Information

¹H and ¹³C NMR spectra of products and other data. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) (a) Plunkett, O.; Sainsbury, M. Rodd's Chemistry of Carbon Compounds, 2nd ed.; 1998; Vol. 4 (Part F/Part G (partial), p 365. (b) Bailey, T. D.; Goe, G. L.; Scriven, E. F. V. Chem. Heterocycl. Compd. 1984, 14 (Pyridine Its Deriv., Pt. 5), 1. (c) Thummel, R. P. Chem. Heterocycl. Compd. 1984, 14 (Pyridine Its Deriv., Pt. 5), 253. (d) Kumar, R.; Chandra, R. Adv. Heterocycl. Chem. 2001, 78, 269. (e) Umeda, H.; Takeuchi, M.; Suyama, K. J. Biol. Chem. 2001, 12579—12587.

(2) (a) Snider, B. B.; Neubert, B. J. Org. Lett. 2005, 7, 2715–2718. (b) Cappelli, A.; Anzini, M.; Vomero, S.; Canullo, L.; Mennuni, L.; Makovec, F.; Doucet, E.; Hamon, M.; Menziani, M. C.; De Benedetti, P. G.; Bruni, G.; Romeo, M. R.; Giorgi, G.; Donati, A. J. Med. Chem. 1999, 42, 1556–1575. (c) Leon, R.; Marco-Contelles, J.; Garcia, A. G.; Villaroya, M. Bioorg. Med. Chem. 2005, 13, 1167–1175. (d) da Costa, J. S.; Pisoni, D. S.; da Silva, C. B.; Petzhold, C. L.; Russowsky, D.;

- Ceschi, M. A. J. Braz. Chem. Soc. 2009, 20, 1448–1454. (e) Pisoni, D. S.; da Costa, J. S.; Gamba, D.; Petzhold, C. L.; de Amorim Borges, A. C.; Ceschi, M. A.; Lunardi, P.; Saraiva Gonçalves, C. A. Eur. J. Med. Chem. 2010, 45, 526–535. (f) Capelli, A.; Anzini, M.; Vomero, S.; De Benedetti, P. G.; Menziani, M. C.; Giorgi, G.; Manzoni, C. J. Med. Chem. 1997, 40, 2910–2921.
- (3) Schrider, M. S. U.S. Patent 4006236A, 1977.
- (4) Messmer, A.; Bátori, S.; Hajós, G.; Benkó, P.; Pallos, L.; Petöcz, L.; Katalin, G.; Kosóczky, I. U.S. Patent 4697013A, 1987.
- (5) (a) Chichibabin, A. E. Bull. Soc. Chim. Fr. 1939, 6 (3), 522–533.
- (b) Edgar, O. B.; Johnson, D. H. J. Chem. Soc. 1958, 3925-3944.
- (c) Chafetz, H.; Anderson, R. C. U.S. Pat. 3349092A, 1967.
- (d) Upadysheva, A. V.; Usova, E. P.; Titova, I. A.; Znamenskaya, A. P. J. Appl. Chem. USSR (Engl. Transl.) 1971, 44, 1127-1132.
- (e) Krishna Mohan, K. V. V.; Narender, N.; Kulkarni, S. J. Microporous Mesoporous Mater. 2007, 106, 229-235.
- (6) Kotsuki, H.; Mehta, B. K.; Yanagisawa, K. Synlett 2001, 8, 1323–1325.
- (7) (a) Chafetz, H.; Patmore, E. L. U.S. Patent 3408351A, 1968.
 (b) Bischoff, V. C.; Herma, H. J. Prakt. Chem. 1976, 318, 891–894.
- (8) (a) Palacios, F.; Alonso, C.; Rubiales, G.; Ezpeleta, J. M. Eur. J. Org. Chem. 2001, 2115–2122. (b) Ruangsiyanand, C.; Rimek, H.-J.; Zymalkowski, F. Chem. Ber. 1970, 103, 2403–2410. (c) Noguchi, M.; Onimura, K.; Isomura, Y.; Kajigaeshi, S. J. Heterocycl. Chem. 1991, 28, 885–890. (d) Upadisheva, A. V.; Grigorieva, N. D.; Znamenskaya, A. P. S.U. Patent 551328A1, 1977.
- (9) Dorofeenko, G. N.; Safaryan, G. P.; Polyakova, T. I. Chem. Heterocycl. Compd. 1972, 8, 1318–1320.
- (10) Driessen, P. B. J.; Grace, D. S. B.; Hogeveen, H.; Jorritsma, H. *Tetrahedron Lett.* **1976**, *17*, 2263–2266.
- (11) Couture, A.; Bochu, C.; Grandclaudon, P. Tetrahedron Lett. 1989, 30, 6865–6866.
- (12) (a) Wotring, L. L.; Townsend, L. B. Cancer Res. 1989, 49, 289—294. (b) Raić-Malić, S.; Grdiša, M.; Pavelic, K.; Mintas, M. Eur. J. Med. Chem. 1999, 34, 405—413. (c) Mischra, R. C.; Dwivedi, N.; Tripathi, R. P.; Bansal, I.; Saxena, J. K. Nucleosides, Nucleotides Nucleic Acids 2005, 24, 15—35.
- (13) (a) Morrey, J. D.; Smee, D. F.; Sidwell, R. W.; Tseng, C. Antiviral Res. 2002, S5, 107–116. (b) Sharma, A. P.; Ollapally, A. P.; Jones, W.; Lemon, T. Nucleosides, Nucleotides Nucleic Acids 1992, 11, 1009–1038. (c) Kabbaj, Y.; Lazrek, H. B.; Barascut, J. L.; Imbach, J. L. Nucleosides, Nucleotides Nucleic Acids 2005, 24, 161–172. (d) Maslen, H. L.; Hughes, D.; Hursthouse, M.; De Clercq, E.; Balzarini, J.; Simons, C. J. Med. Chem. 2004, 47, 5482–5491. (e) Rusinov, V. L.; Egorov, I. N.; Chupakhin, O. N.; Belanov, E. F.; Bormotov, N. I.; Serova, O. A. Pharm. Chem. J. 2012, 45, 655–659; Engl. Transl. Khim.-Farm. Zh. 2011, 45 (11), 7–11.
- (14) (a) Khalil, N. S. A. M.; Mansour, A. K.; Eid, M. M. Nucleosides, Nucleotides Nucleic Acids 2004, 23, 1889–1910. (b) Modzelewska-Banachiewicz, B.; Kaminska, T. Eur. J. Med. Chem. 2001, 36, 93–99. (15) Rusinov, V. L.; Zyryanov, G. V.; Pilitcheva, T. L.; Chupakhin, O. N.; Neunhoeffer, H. J. Heterocycl. Chem. 1997, 34, 1013–1019.
- (16) (a) Chupakhin, O. N.; Rusinov, G. L.; Beresnev, D. G.; Neunhoeffer, H. J. Heterocycl. Chem. 1997, 34, 573–578. (b) Rusinov, G. L.; Beresnev, D. G.; Itsikson, N. A.; Chupakhin, O. N. Heterocycles 2001, 55, 2349–2360. (c) Chupakhin, O. N.; Rusinov, G. L.; Itsikson, N. A.; Beresnev, D. G.; Fedorova, O. V.; Ovchinnikova, I. G. Russ. Chem. Bull. 2004, 53, 2308–2313; Engl. Transl. Izv. Akad. Nauk, Ser. Khim. 2004, 2210–2215. (d) Beresnev, D. G.; Itsikson, N. A.; Chupakhin, O. N.; Charushin, V. N.; Kodess, M. I.; Butakov, A. I.; Rusinov, G. L.; Morzherin, Yu.Yu.; Konovalov, A. I.; Antipin, I. S. J. Org. Chem. 2006, 71, 8272–8275.
- (17) Egorov, I. N.; Kovalev, I. S.; Rusinov, V. L.; Chupakhin, O. N. Z. *Naturforsch.* **2010**, 65b, 1359–1362.
- (18) Weiss, M. J. Am. Chem. Soc. 1952, 74, 200-202.
- (19) Klumpp, D. A.; Garza, M.; Jones, A.; Mendoza, S. J. Org. Chem. 1999, 64, 6702–6705.
- (20) (a) Yadav, J. S.; Reddy, B. V. S.; Gupta, M. K.; Prathap, I.; Dash, U. *Synthesis* **2007**, 1077–1081. (b) Nikam, S. S.; Sahasrabudhe, A. D.;

- Shastri, R. K.; Ramanathan, S. Synthesis 1983, 145–147. (c) Hamana, M.; Iwasaki, G.; Saeki, S. Heterocycles 1982, 17, 177–181.
- (21) Uchutilova, V.; Fiedler, P.; Prystas, M.; Gut, J. Collect. Czech. Chem. Commun. 1971, 36, 1955–1963.
- (22) Heinisch, L. J. Prakt. Chem. 1974, 316, 667-678.
- (23) Iogansen, A. V.; Kurkchi, G. A.; Baeva, V. P.; Rasskazova, Z. N.; Salamatina, G. A. Russ. J. Org. Chem. 1971, 7, 2608–2610; Engl. Transl. Zh. Org. Khim. 1971, 7, 2509–2511.
- (24) Kelly, K. K.; Matthews, J. S. J. Chem. Eng. Data 1969, 14, 276-277.
- (25) Martin, A. U.S. Patent 5776884, 1998.